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Calculate ΔH for the
process. Use $q = mC\Delta t$

$\text{KCl(s)} + \text{K}^+(\text{aq}) +$

$\text{Cl}^-(\text{aq}) \Delta H = ?$ $q = (65.0$

$\text{g})(4.184 \text{ J})(-10.5^\circ\text{C})$ g

$^\circ\text{C}$ $q = -2855.58 \text{ J}$

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C105 Student Self-

evaluation Chapter 6

Thermochemistry Upon

finishing this chapter,

students will be able to

(student learning

outcomes): 1. Define

energy, heat, work,

internal energy,

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internal energy
change, enthalpy and
enthalpy change and
classify these terms as
state function or non-
state function. 2.

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Chemistry 1 Chapter 5

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Objectives: * Discuss

what is studied in the

subject of

thermochemistry and

why this subject is

relevant * State the

First Law of

Thermodynamics and

explain why it is

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important and how it is applied * Identify

endothermic and

exothermic processes *

Calculate the amount

of heat energy

transferred based on a

given amount of a

substance in a

chemical reaction with

a known enthalpy of

reaction (ΔH) *

Calculate ΔH for a target

reaction using ...

CHM4220 Chapter 5

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Chemistry 1...

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studying

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Thermochemistry $Q =$

$m C \Delta T$ $\Delta H = -q$ $C =$

4.184 J moles g oC 1.

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Between Ionic and
Molecular.... Heat
transfers from objects
of higher temperature
to objects o.... Higher
temperature, fast
moving, particles
collide with lower....

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Quizlet**

As you study
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Thermochemistry, you will learn how different forms of energy apply to Chemistry, such as Transferring Heat Energy: Exothermic and Endothermic and the Definition of Catalyst. You will learn about laws of energy that are important to the study of Thermochemistry: 1st Law: Conservation of Energy and 2nd Law: Entropy.

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Student Study Guide Thermochemistry

The guide is organized similarly to the general chemistry guide with a clear separation of first-term and second-term material. Each chapter is organized with study and practice questions

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Thermochemistry

where the study questions take you through the problem solving process of key problems explaining the correct process and also explaining the incorrect processes leading to incorrect answers.

Student Study Materials | ACS Exams

While on the topic of teaching thermochemistry, it is

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worth flagging a useful
review of the research

on teaching and
learning early

undergraduate

thermodynamics

published recently by

Towns et al in

Chemistry Education

Research and Practice

(CERP - free to access).

2 This covers various

aspects of teaching

thermodynamics: the

role of maths, students

view of matter, and a

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conceptions.

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Lessons ...

Thermochemistry is the scientific study of the heat energy that is involved in or produced by chemical reactions and/or physical transformations. What can you tell us about how this topic is studied and what its uses are?

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**Thermochemistry
Quizzes Online,
Trivia, Questions ...**

Thermochemistry

Energy changes that accompany chemical reactions will be studied. We will be able to calculate the amount of heat gained or lost as the chemical bonds are broken and rearranged as matter undergoes chemical changes. PE Diagram
Video:

Exo/endothermic rxns

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Hess's Law Gibbs Free
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