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Balancing Chemical Equations Worksheet Answers

If you also get perplexed in balancing chemical equations, follow the tips for

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correct balancing
chemical equations
worksheet answers. Tip
1: When you are
trying to balance the
chemical equations,
you should remember
that you can only
change the value of
coefficient in front of
the element or
compound, and not the
subscript.

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Before you start balancing chemical equations, it is important that you become firmly acquainted with the various part of one. Every balanced chemical equation consists of two parts: the reactant side and the product side. Both of these sides are separated by the means of an arrow. On the left side of the

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arrow, you will find the reactant side.

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Key Balance the

equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightleftharpoons 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightleftharpoons 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightleftharpoons 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightleftharpoons 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightleftharpoons 2$

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H₂O + 1 PbCl₂ 6) 2
AlBr₃ + 3 K₂SO₄ Æ 6
KBr + 1 Al₂(SO₄)₃ 7) 1
CH₄ + 2 O₂ Æ 1 CO₂ +
2 H₂O 8) 1 C₃H₈ + 5
O₂ Æ 3 CO₂ + 4 H₂O
9) 2 C₈H₁₈ ...

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Balancing Chemical Equations Answer Sheet Worksheets ...

Balance the equations below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$. 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$.

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Balancing Chemical Equations Worksheet 2

- ANSWERS
26. $2\text{Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$
27. $2\text{Ag} + \text{O}_2 \rightarrow 4\text{Ag} + \text{O}_2$
28. $4\text{K} + 2\text{O}_2 \rightarrow 2\text{K}_2\text{O}$
29. $\text{Cl}_2 + 3\text{F}_2 \rightarrow 2\text{ClF}_3$
30. $\text{SiO}_2 + 2\text{C} \rightarrow \text{Si} + 2\text{CO}$
31. $2\text{NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{CO}_2 + \text{H}_2\text{O}$
32. $2\text{ZnS} + 3\text{O}_2 \rightarrow 2\text{ZnO} + 2\text{SO}_2$
33. $\text{SO}_2 + 2\text{H}_2\text{S} \rightarrow 3\text{S} + 2\text{H}_2\text{O}$
34. $2\text{Ba} + \text{O}_2 \rightarrow 2\text{BaO}$
35. $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
36. $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$
- 37.

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**Balancing Chemical
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unbalanced equation is
mostly a matter of
making certain mass
and charge are

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balanced on the reactants and products side of the reaction arrow. This is a collection of printable worksheets to practice balancing equations. The printable worksheets are provided in pdf format with separate answer keys.

How to Balance Equations - Printable Worksheets

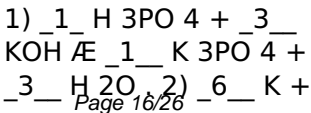
Balancing Equations:

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Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).

Balancing Equations: Practice Problems

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**Balancing Chemical
Equations Practice
Worksheet With ...**

Worksheet: Writing and
Balancing Chemical
Reactions 1. Balance
the following equations
and indicate the type
of reaction as
formation,
decomposition, single
replacement, double
replacement,
hydrocarbon
combustion, or other.

... Answers Question 1

a. 2Cu (s)

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Worksheet: Writing and Balancing Chemical Reactions

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Student Instructions for
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Identify the reactants
and products and write
a word equation. Write
the correct chemical
formula for each of the
reactants and the
products. Check to see
whether there are the
same NUMBERS and
TYPES of atoms on
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